

Mineral-Water Reaction at High Pressures – Implications for Uranus and Neptune. S.-H. Dan Shim¹, C. Nisr¹, T. Kim², Y. J. Lee², A. Chizmeshya¹, K. Leinenweber¹, S. Chariton³, V. Prakapenka³, S. Speziale⁴, Z. Liu⁵, and H.-P. Liermann⁶. ¹Arizona State University (781E Terrace Mall, Tempe, AZ 85287; shdshim@asu.edu), ²Yonsei University, ³University of Chicago, ⁴GFZ German Research Center for Geosciences, ⁵University of Illinois at Chicago, and ⁶Deutsches Elektronen Synchrotron.

Introduction: Uranus and Neptune have a thick H₂O-rich layer above the core (rocks and metals). Although the standard models have assumed no mixing between these materials, recent models have invoked significant amounts of heavy elements in the H₂O-rich layer with a compositional gradient to explain very little heat flow from the interior of Uranus [1, 2]. However, little is known about how heavy elements can exist there with long-term dynamic stability. Processes which can mix heavy elements in H₂O-rich layer are also important for understanding water-rich exoplanetary bodies [3]. We have conducted a series of experiments and density functional theory calculations to understand reaction between minerals and H₂O at high pressures and temperatures expected for the interiors of Uranus and Neptune. We also discuss implications of the experimental observations.

Methods: For high-pressure experiments, we loaded pure mineral phases together with H₂O in diamond-anvil cells. The mineral phases we studied are (Mg,Fe)₂SiO₄ olivine, (Mg,Fe)O ferropericlae, and SiO₂. The samples were laser heated to 700-2000 K in a H₂O medium at 7-110 GPa. X-ray diffraction patterns were measured in situ at high pressures and temperatures at the GSECARS sector of Advanced Photon Source. Infrared spectra of the samples were measured at Brookhaven National Lab. The recovered samples have been analyzed for chemical composition and morphology at Arizona State University and Yonsei University. We also have conducted density functional theory calculations (DFT).

Results: We found that olivine reacts with H₂O at high pressure and high temperature. The texture and chemical composition of the reaction products indicate that MgO is leached out from olivine and dissolved in H₂O. The solubility of MgO in H₂O peaks at 20-40 GPa. At pressure, the MgO solubility is comparable to the solubility of NaCl in H₂O at ambient conditions. SiO₂ component remains solid. Separate experiments showed that SiO₂ can contain a large amount of H₂O in the crystal structure at high pressure: $x = 0.2$ in (Si_{1-x}H_{4x})O₂. H₂O also alters the phase behavior of SiO₂, stabilizing NiAs-type structure at pressures above 60 GPa. The phase is not thermodynamically stable in dry SiO₂.

Discussion and Implications: Our observations found a significant amount of dissolved MgO in H₂O at high pressures and temperatures relevant to water-

rich planets. The experimentally observed behavior of MgO could result in a compositional gradient in the outer layer of Uranus. After peaking at 20-40 GPa, the solubility would decrease with depth according to our experiments. The selective dissolution of MgO in H₂O could result in silica-rich rocky core. The large amount of H₂O dissolved in crystal structure of SiO₂ solid polymorphs suggest that the rocky core of water-rich planet could be extensive hydrated (Fig. 1). It has been believed that the enhanced conductivity of H₂O through a series of phase changes play an important role for the observed magnetic field of Uranus and Neptune [4]. If a significant amount of MgO can be dissolved in the H₂O layer, MgO could affect the properties of the layer. It is also feasible that the H₂O and some heavy elements may experience mixing and de-mixing at different depths in the interiors of Uranus and Neptune, affecting the geochemical cycle.

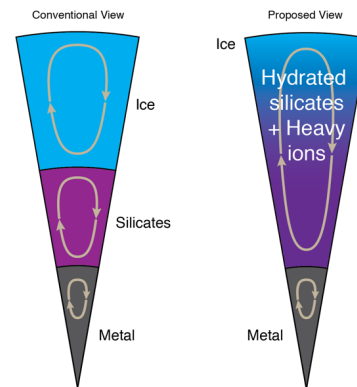


Figure 1: The internal structure of the water-rich planets. (Left) Conventional view. (Right) Proposed structure based on inter-solubility of rock and ice.

Acknowledgements: The work has been supported by NASA (80NSSC18K0353) and NSF (EAR1338810). The results reported herein benefit from NASA's Nexus for Exoplanet System Science (NEXSS) research coordination network.

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