

**POWDER X-RAY DIFFRACTOMETRY OF THE ORGUEIL CARBONACEOUS CHONDRITE: INSIGHTS INTO THE CLAY MINERALOGY** Laurence A.J. Garvie<sup>1,2</sup>, <sup>1</sup>Center for Meteorite Studies, <sup>2</sup>School of Earth and Space Exploration, Arizona State University, 781 East Terrace Rd., Tempe, AZ 85287-6004 (lgarvie@asu.edu).

**Introduction:** The Orgueil carbonaceous chondrite has received considerable scientific attention since its fall in 1864 [1], in part because of its cosmic composition and hydrous clay-rich nature. The identity of the clay has proven difficult in part because of its fine-grained nature and broad, relatively indistinct reflections. The dominant  $\sim 7$  and  $14 \text{ \AA}$  reflections have been ascribed to a range of phyllosilicates. However, the reflections are most consistent with serpentine and smectite [2]. TEM studies show compositional data and d-spacings consistent with Fe-bearing, Mg-rich serpentine and saponite [3]. In addition, the HRTEM images suggest a certain degree of interstratification of serpentine and saponite. Quantitative powder XRD show that Orgueil is  $\sim 83$  vol% serpentine/saponite [4]. Here, further insights into the bulk nature of the clays are provided by powder x-ray diffraction (XRD) of samples prepared using ‘standard’ clay identification procedures.

**Samples and preparation:** Powder XRD patterns were acquired with a Rigaku MiniFlex 600 diffractometer, with a post-diffraction monochromator, and Cu K $\alpha$  radiation. Data were acquired from whole rock and separated clays fractions. Clay fractions were prepared by crushing  $\sim 0.5$  g of Orgueil in water and using a centrifuge to separate the 2 to 0.5, and  $<0.5$ - $\mu\text{m}$  clay fractions. These fractions were then divided into two portions and exchanged with 0.1M CaCl<sub>2</sub> and 1M KCl solutions, respectively. These clays were then mixed with a few ml of water and deposited onto glass slides forming oriented clay mounts. XRD patterns were acquired under a range of humidities, after saturation with ethylene glycol (60 °C, 24 hrs), and after heating at 300 ° and 500 °C for 1 hour.

**Results and discussion:** The bulk powder XRD profiles show sharp, intense reflections for magnetite, and less intense sharp peaks for dolomite, pyrrhotite, and possibly olivine. These peaks sit on a series of broad reflections from clays, prominently observed by the broad low-angle peaks near  $15.5$  and  $7.5 \text{ \AA}$ , and two prism-like structures starting near  $20^\circ$  and  $25^\circ 2\theta$ . The overall pattern is similar to that acquired by synchrotron XRD [Nozaki], with minor differences in the intensities of the sharp reflections.

The air-dried, Ca-saturated,  $<0.5 \mu\text{m}$  clay fraction of randomly oriented and oriented samples show intense but broad reflections centered near  $15.5$  and  $7.5 \text{ \AA}$ , and a prism-structure near  $25^\circ 2\theta$  (Fig. A,B). After K-saturation the  $15.5 \text{ \AA}$  reflection is broader and shifts to  $12.8 \text{ \AA}$ . The randomly oriented pattern from the air-dried Ca-saturated clay is dominated by 00 $l$  reflections for smectite and serpentine and two-dimensional  $hk$  diffraction bands from smectite (Fig. A). The smectite prism is absent in the patterns from the oriented clay samples, though the serpentine prism is prominent.

Ethylene-glycol solvation shifts the 001 peak from the Ca-saturated smectite to  $18.06 \text{ \AA}$  (Fig. B), with a

new peak at  $8.34 \text{ \AA}$ . The K-saturated clay (not shown) has a broad, poorly resolved peak at  $19.7 \text{ \AA}$  and less intense reflection at  $8.39 \text{ \AA}$ . After heating the Ca-saturated clay to  $300^\circ$  and  $500^\circ \text{C}$ , the 001 reflection collapses to  $9.76$  and  $9.32 \text{ \AA}$ , respectively. For all of the treatments, a comparatively weak intensity 001 serpentine peak remains unshifted at  $7.46 \text{ \AA}$ .

Ferrihydrite has been identified by electron diffraction in Orgueil [3] and said to be ‘very abundant’, with reflections at  $\sim 2.58$  and  $1.50 \text{ \AA}$ . However, these reflections overlap with the smectite  $hk$  diffraction bands: hence ferrihydrite cannot be definitively identified in the clay patterns, though suffice it to say that it does not contribute significant intensity to the patterns.

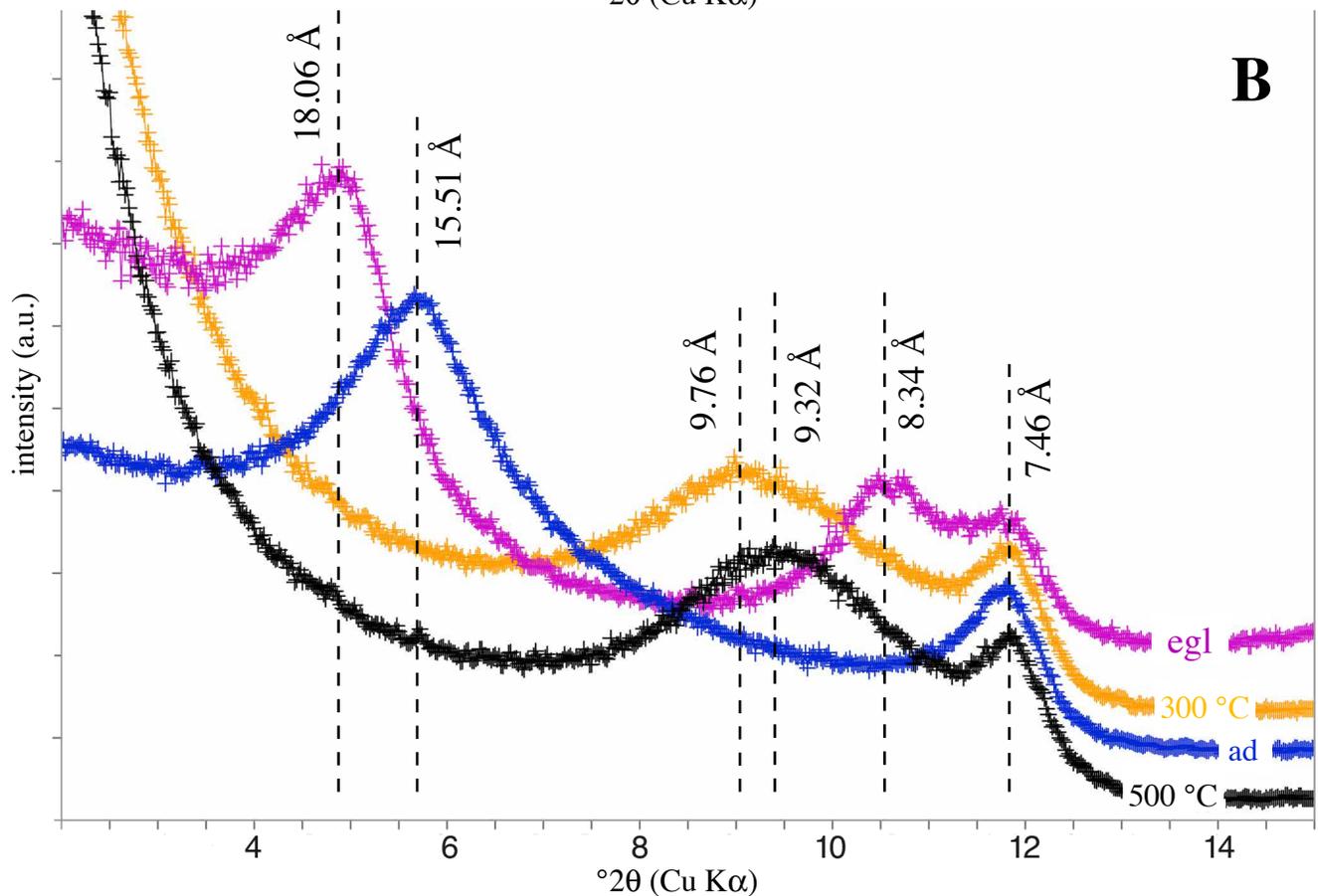
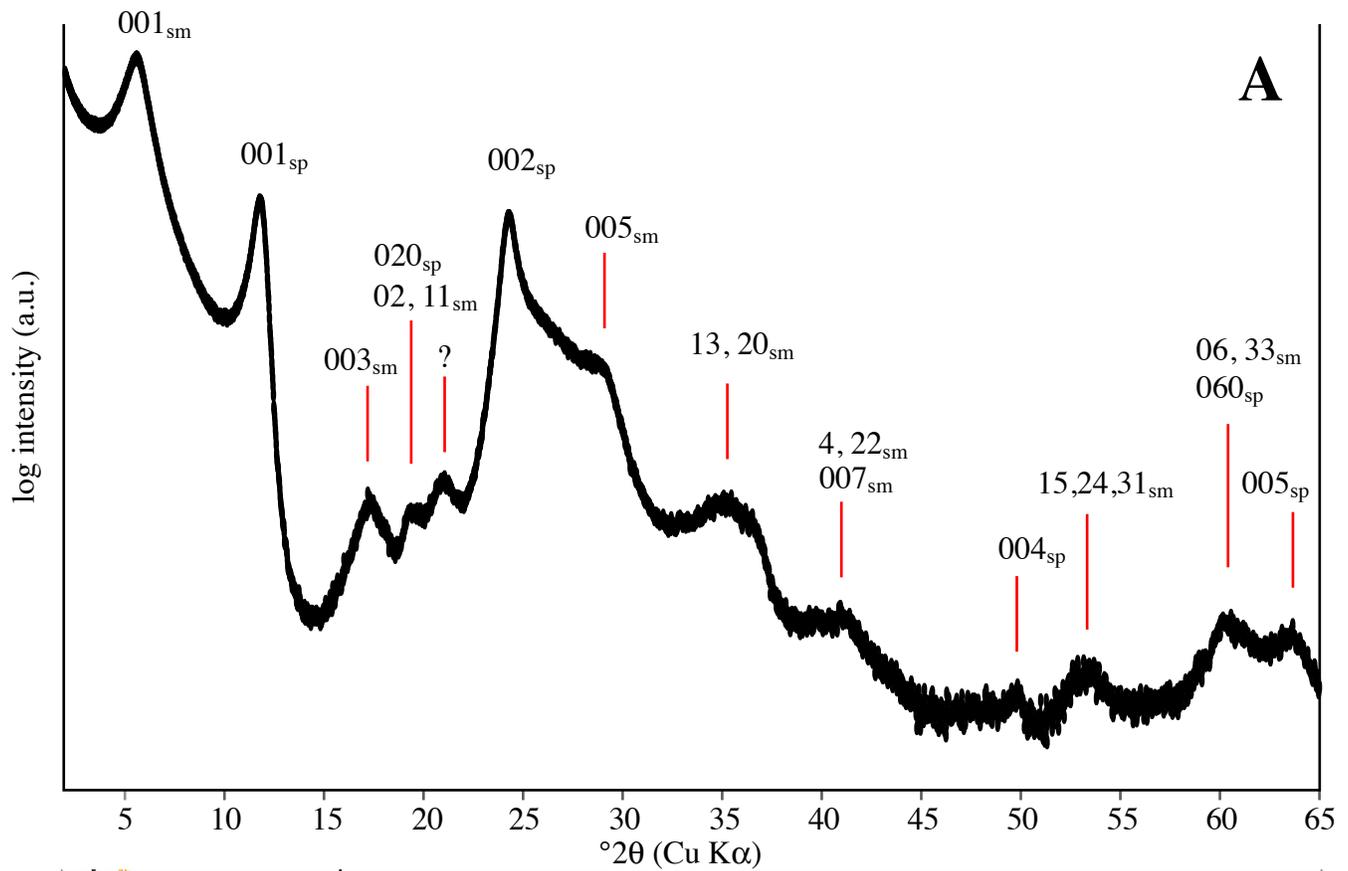
The d-spacings and behavior of the reflections after the treatments are broadly consistent with a mixture of serpentine and smectite, consistent with previous analyses [2]. However, the serpentine 001 spacing of  $7.46 \text{ \AA}$  is significantly larger than that observed in other serpentines, which are typically  $7.2$  to  $7.3 \text{ \AA}$ . The d-spacing of the 06, 33  $kh/060$  reflection match that for trioctahedral clays. Ethylene-glycol-solvated smectites typically yield a basal spacing near  $\sim 17 \text{ \AA}$ , with a rational series of higher-order reflections. In contrast, the Orgueil clay swells to  $\sim 18.06$  to  $19.7 \text{ \AA}$ .

The patterns are consistent with clays that are fine-grained and exhibit a high degree of disorder. While TEM has revealed well-crystallized grains, the XRD patterns indicate that disordered clays predominate. The broadness of the smectite peaks is a result of a combination of interstratification, intercalation, and small grain size. Intercalation, such as from silica [6] or carbon [7], affects the swelling properties and could partly account for the broadness of the reflections after heating. The non-integer 001 versus 002 spacing could indicate a degree of random interstratification. The relatively high 001 d-spacing for the serpentine is likely the result of small diffracting domains with, on average, few  $\sim N < 10$  layers per crystallite.

A question the XRD data raises is why aren’t the clays better crystallized? As reviewed in [1], alteration occurred between  $0^\circ$  and  $150^\circ \text{C}$ , and assuming the clay formation coincided with the crystallization of the dolomite and breunnerites, then formation occurred over  $\sim 7$  to  $9$  Myrs time period. The poorly crystalline nature of the clays suggest they formed rapidly and did not undergo significant recrystallization by Ostwald ripening.

**Acknowledgement:** Partial support for this research was provided by the NASA Emerging Worlds (EW) program through grant NNX17AE56G.

**References:** [1] Gounelle and Zolensky (2014) MAPS, 49, 1769-1794. [2] Bass (1971) GCA, 35, 139-147. [3] Tomeoka and Buseck (1988) GCA, 52, 1627-1640. [4] King et al. (2015) GCA, 165, 148-160. [5] Nozaki et al. (2006) MAPS, 41, 1095-1114. [5] Endo et al. (1980) Clays Clay Miner, 28, 105-110. [6] Garvie and Buseck (2007) MAPS, 42, 2111-2117.



Powder X-ray diffraction (XRD) data from the  $<0.5 \mu\text{m}$  Ca-saturated Orgueil clay fraction. A) Pattern from a randomly oriented mount under air-dried conditions. sm - smectite, sp - serpentine. B) Patterns from oriented mounts acquired under the air dried (ad), ethylene glycol saturated (egl), and heated to 300 and 500 °C conditions. Profiles arbitrarily shifted along the y-axis for clarity.