

SECONDARY-VOLATILES LINKED METALLIC IRON IN EUCRITES: THE COMPLEX CASE OF CAMEL DONGA. Paul H. Warren¹ and Junko Isa¹, ¹Dept. Earth, Planetary & Space Sci., UCLA, Los Angeles, CA 90095-1567, USA; pwarren@ucla.edu.

The recent discovery of enigmatic large, vein-shaped masses of native Fe in NWA 5738, a eucrite with an extraordinarily diverse array of secondary-metasomatic alterations [1], has prompted interest in other Fe-metal rich eucrites. As described by Cleverly et al. [2] and [3], Camel Donga contains about 2.1 wt% (1 vol%) metallic iron. For the most part, this metal occurs as scattered small (<0.1 mm) grains of high-purity Fe composition: <0.05 wt% Ni according to [2], 0.058 wt% according to an INAA study of aggregated particles [3]. In our own study, from a typical sample (CD1), out of 35 scattered points analysed, 24 yielded results below our detection limit of 0.012 wt% (using high current and long count durations), and the other 11 average 0.03 wt%; the only consistently detectable impurity, Co, averages (N=35) 0.036 wt%.

Palme et al. [3] developed an elegant model for origin of these pure-Fe metals by reduction, with most of the Fe deriving from reduction of the ferrosilite component of the rock's pyroxene. These authors noted that bulk-rock data [cf. 2, 4] show no increase in Fe over the normal concentration (~144 mg/g) for a ferroan noncumulate eucrite, so they inferred that iron was reduced in situ, not added, by the secondary processing of this rock. As reducing agent, they suggested that S₂ gas formed by decomposition of troilite (FeS) in a reheating episode, and then after combining with pyroxene-derived O, left the rock as SO₂ gas.

However, this simple pyroxene-reduction, SO₂-loss model does not fit so well with more recent observations on Camel Donga. Two bulk analyses by [5] found higher Fe, 150 and 177 mg/g, and the latter result (significantly enriched vs. most noncumulate eucrites) came from a powder produced from an exceptionally large (1.8 g) mass of the meteorite. Also in contrast with earlier studies [2, 3], we have found that the compositional range for low-Ca pyroxene (again, using our "typical" sample CD1) extends up to *mg* = 51 mol%. This spread is somewhat enigmatic, because for the most part our results fall along a tie line between En₄₁Wo₂ opx and En₃₂Wo₄₃ augite [cf. 2, 3]. Still, the higher *mg* results imply that the rock's bulk-silicate Fe must be slightly lower than previously supposed.

We have also studied an additional sample of Camel Donga (CD2; unfortunately severely weathered) that is atypical inasmuch as it contains an 11-mm, sharp-bounded, ovoid or lens-shaped nodule with vastly higher original metal content than the surrounding material (Fig. 1; nodule is rust-colored; black is fusion crust). The metal has mostly been altered to oxide, but

the other phases in the nodule were not so greatly affected by the weathering. The nodule's pyroxenes consistently have compositions along the aforementioned, typical-CD tie line (e.g., En₄₁Wo₂ opx); no extraordinary pyroxene reduction is evident in association with the extraordinary-high metal content.



The CD2 nodule's metal, and also the metals in a smaller tight cluster nearby (3 mm outside the nodule), have distinctively Ni-enriched compositions. For the nodule, 21 analyses avg. 0.525±0.009 wt% Ni and 0.040±0.007 wt% Co. For the nearby cluster, 5 analyses give practically identical averages. For 9 analyses of metals in the same CD2 thin section but far from the nodule, the averages are 0.064 wt% Ni and 0.044 wt% Co; more similar, although still not quite so pure-Fe, compared to our CD1 results (see above). Also, one of the bulk-rock analyses [4] shows Ni enrichment (38 µg/g). Such high Ni contents would be unprecedented in a eucrite free of impactor-debris "contamination", and thus imply that not all of the Fe-metal in Camel Donga formed by simple in-situ reduction.

The opaque oxides of Camel Donga (CD1) show surprisingly little evidence for involvement in the in-situ reduction process. Margins as well as interiors of ilmenites show Fe-metal no more abundant than in the meteorite as a whole, and no rutile has yet been found. The spinels also appear unaffected, including in terms of *mg*, even at their rims. None of the Mg- and Cr-rich ilmenite that can form when spinel is reduced [6] has been found in Camel Donga. It seems mysterious that a brief, high-*T* redox episode would attack pyroxene while leaving FeO-rich oxides unmolested. Diffusion is much faster in oxides than in pyroxene [7-9].

In CD1, an aspect of similarity to NWA 5738 [1] was found in the form of a discontinuous, curvy vein of

anomalously Na-poor (An_{96}) plagioclase within one of the larger pyroxenes. We suspect that Camel Donga is a product of a complex sequence of alterations that included, and may have begun with, localized mechanical admixture of impactor-debris “contamination”. In modeling NWA 5738, [1] noted that Dawn results have shown that the crust of Vesta (and by extension, for that matter, any basalt-crust belt asteroid) appears to be remarkably rich in volatile-rich carbonaceous-chondritic matter that remains relatively intact through the accretion process (which is gentle, in this context, compared to more familiar Earth and Moon settings). Fluids emanating from such masses, when heated, are likely to be diverse initially and then evolve to even greater diversity, including redox diversity. Camel Donga may represent one, most highly reduced, end of a spectrum of eucrite secondary alteration scenarios that also includes extraordinarily oxidized materials, as found from spinel (and olivine) analyses for the most common variety of alteration veins in NWA 5738 [1].

Meanwhile, we have also studied (again using the stoichiometric methodology of [1]) spinel-olivine couplings in alteration veins of the Haraiya [10] and NWA 5073 [11] eucrites. Results indicate uncommonly high oxidation states (Fe^{3+} in spinel), by eucrite standards, although not so high as in NWA 5738. The link between such oxidized materials and sulfurized eucrites such as NWA 2339 [12], and to the troilite-decomposition mechanism proposed by [2], should be the subject of exciting further research.

Acknowledgements: We thank the Smithsonian Institution and the University of Münster (Addi Bischoff) for the loan of Haraiya and NWA 5073 samples, respectively.

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