## I-Xe RECORD OF AQUEOUS ALTERATION IN CK MAGNETITES.

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**Introduction:** Magnetite is one of the first products formed in carbonaceous chondrites during aqueous alteration, providing a valuable timeline for metamorphic changes on carbonaceous chondrite parent bodies [1]. Based on the I-Xe ages of CI Orgueil, CM Murchison and CVs Bali, Kaba, Mokoia and Vigarano [2, 3], aqueous alteration in carbonaceous chondrites started early, ~ 3.5 Ma after formation of CV CAIs [4].

Here we present data for magnetites separated from the CK chondrites of different metamorphic grades.

**Experimental:** Eight CK meteorites were selected for this study (Table 1). The majority experienced S2 degree shock; the degree of weathering warried from W1 to W3. The samples were finely ground and stirred with a saturated NaOH solution for 8 days at 60°C. This procedure demonstrated to yield magnetic fractions that are at least 90% pure [5]. The resulting magnetite separates and the absolute age standard Shallowater aubrite [6] were irradiated with  $2 \times 10^{19}$  thermal neutrons/cm<sup>2</sup> to convert <sup>127</sup>I into <sup>128</sup>Xe. Xe was released in 17-18 temperature steps; its isotopic composition was measured using high-transmission mass-spectrometry.

СК	Туре	W, S	Mag, %	I-Xe age, Ma	<sup>i</sup> Xe×10 <sup>-10</sup> , cm <sup>3</sup> STP/g		
Metorite					<sup>129*</sup> Xe	<sup>132</sup> Xe <sub>tr</sub>	<sup>132</sup> Xe <sub>fs</sub>
Hart	3	W2/3, S2	2.6	—	0.52	12.93	0.21
NWA 6047	3	W3, S2	-				
NWA 1559	3	W2, S2	3.9	$0.2\pm0.5$	4.60	3.16	1.12
NWA 5956	3	W1, S2	1.2	—	0.21	3.92	0.03
NWA 5343	3		4.5	$0.6 \pm 0.5$	1.70	1.08	0.10
NWA 5515	4	md, min	15.8	$1.4 \pm 1.2$	1.47	0.64	0.05
NWA 8672	5	Low, S2	9.3	$0.8\pm0.5$	2.39	0.87	0.06
NWA 8670	6	Low, S2	8.6	-	0.46	0.27	0.07

## Table 1.

I-Xe ages and concentrations of Xe components in magnetites, separated from CK chondrites (tr – trapped; fs – U-fission; \* – I-derived; <sup>†</sup>- moderate; <sup>††</sup>- minimal).

**Results:** Only CK3 NWA 6047 failed to yield magnetite separate; for the metamorphic grade 4-6 samples, the amounts of separated magnetite is higher than those for the CK3s. Concentration of trapped <sup>132</sup>Xe in the analyzed samples appears to decrease with increasing metamorphic grade, although it varies by more than order of magnitude for CK3s. CK3s Hart and NWA5956 lost more than 90% of their <sup>129</sup>\*Xe, most likely due to a shock experienced

Shallowater CI1 Orgueil [2] <u></u> 1568.2 ± 0.16 (Pb-Pb age of CAls CM2 Murchison [3] Bali CV3 Kaba [3] Mokoia Vigarano СКЗ NWA 1559 СКЗ NWA 5343 СК4 NWA5515 CK4 Karoonda [5 CK5 NWA 8672 -5 -4 -3 -2 2 3 -1 0 Relative I-Xe age, Ma

**Figure 1.** I-Xe ages (relative to Shallowater) of magnetites separated from carbonaceous chondrites of different types.

after decay of <sup>129</sup>I. CK6 NWA 8670 failed to yield an isochron since the release profiles of I-derived <sup>128\*,129\*</sup>Xe and trapped <sup>132</sup>Xe correlate. This could be due to the metamorphic processes that lead to the loss and subsequent redistribution of the trapped and I-derived Xe in this largely recrystallized CK6. Four CK magnetites provided well-defined isochrones (Figure 1, red symbols), corresponding to the I-Xe ages that agree within the uncertainties. These ages are also in agreement with the previously reported I-Xe age of CK4 Karoonda [5]. It seems that the I-Xe system in these samples recorded the onset of aqueous alteration and was not affected by thermal metamorphism up to degree 6.

I-Xe ages of CKs studied here indicate that CV3s and CKs experienced alteration almost simultaneously, consistent with the single stratified parent body origin.

**References:** [1] Krot A. N. et al. 2006. Meteorites and the early solar system II (eds. Lauretta & McSween, Jr.). The Univ. of Arizona press. [2] Pravdivtseva O. et al. 2018. *Geochimica et Cosmochimica Acta* 227:38–47. [3] Pravdivtseva O. et al. (2013) *LPS XLIV*, Abstract #3104. [4] Connelly J. N. et al. 2012. *Science* 338:651–655. [5] Lewis R. S. & Anders E. 1975. *Proc. National Academy of Sci.* 72:268–273. [6] Pravdivtseva O. et al. 2017. *Geochimica et Cosmochimica Acta* 201:320–330.