THE RELATIONSHIP BETWEEN OXYGEN AND MAGNESIUM ISOTOPE RATIOS IN OLIVINE FROM THE COMET 81P/WILD 2: A COMPARISON WITH AOAS IN PRIMITIVE METEORITES. K. Fukuda¹, D. J. Joswiak², D. E. Brownlee², B. L. Beard¹, M. J. Spicuzza¹, T. J. Tenner³, M. Kimura⁴, N. T. Kita¹.

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Introduction: Comets are thought to have formed at the cold, outer parts of the solar system, providing a different perspective on conditions of early solar system evolution, relative to meteorites. Comet 81P/Wild 2 samples returned by the NASA Stardust mission contain ¹⁶O-rich, FeO-poor, Mn-enriched (LIME) olivine particles that are considered to be condensates from solar nebula gas [1-5]. LIME olivine is also found in amoeboid olivine aggregates (AOAs) in chondrites [e.g., 6, 7]. AOAs in the least metamorphosed meteorites are ¹⁶O-rich [e.g., 8, 9], which suggests a genetic relationship between Wild 2 LIME olivine grains and AOAs. Some AOAs and Ca, Alrich inclusions (CAIs) have negative δ^{25} Mg values [e.g., 10-12] relative to the Earth and bulk chondrites $(\delta^{25}\text{Mg}_{DSM3} = -0.13 [13])$. The light isotope enrichments in these refractory inclusions may be the result of condensation from solar nebula gas due to kinetic isotope fractionation [14].

If ^{16}O -rich Wild 2 olivines are condensates from solar nebula gas, they may have negative $\delta^{25}\text{Mg}$ values. Here we report for the first time Mg isotope ratios of Wild 2 olivine particles with known oxygen isotope ratios from previous studies [2-3, 15]. We also measured oxygen and magnesium isotope ratios in AOAs from the Kaba (CV3.1) and DOM 08006 (CO3.01) chondrites in order to compare with the newly obtained Wild 2 dataset.

Samples and analytical procedures: Seven Wild 2 olivine particles were analyzed from four different tracks (T57, T77, T149, and T175). Mg#'s of the olivine particles range from 60 to 99.8 [1-3, 15]. Previous studies revealed that four out of seven particles are $^{16}\text{O-rich}$ (T57/F10, T77/F6, T175/F1, and T77/F50; $\delta^{18}\text{O:}$ –56 to –46‰; $\Delta^{17}\text{O:}$ –24 to –22‰) and the other three particles are $^{16}\text{O-poor}$ (T77/F4, T149/F1, and T149/F11a; $\delta^{18}\text{O:}$ 2.5 to 6.8‰; $\Delta^{17}\text{O:}$ –1.7 to –1.0‰) [2-3, 15]. We studied one Kaba AOA that is a dense object with little porosity (referred to as "compact" by [16]) and eight DOM 08006 AOAs with compact and/or more porous textures.

Magnesium three-isotope analyses were performed using the WiscSIMS Cameca IMS 1280 equipped with a radio-frequency (RF) plasma ion source. We used O₂⁻ primary ions instead of O⁻ to achieve higher

secondary ion yield and better ionization efficiency [17]. A 2 μ m diameter (25 pA) primary ion beam was used for Wild 2 olivine analyses. A 5 μ m or 9 μ m diameter (160 pA or 2 nA) primary beam was used for AOA olivine analyses. Secondary Mg ions (24 Mg⁺, 25 Mg⁺, 26 Mg⁺) were detected on multicollection Faraday Cups (FCs) using three 10^{12} ohm resistors (25 pA and 160 pA) or one 10^{10} ohm and two 10^{11} ohm resistors for 24 Mg⁺, and 25,26 Mg⁺, respectively (2 nA). The typical external reproducibility (2 standard deviation: 2SD) of δ^{25} Mg for the San Carlos olivine (SC-Ol) standards was 0.25‰, 0.13‰, and 0.08‰ for primary ion beams of 25 pA, 160 pA, and 2 nA, respectively.

We prepared 13 olivine standards with a range of Mg# (100-60) for SIMS instrumental bias corrections. Mg isotope ratios in DSM-3 scale of three standards were previously obtained using solution MC-ICPMS [18]. Including these three, all 13 olivine standards were measured using a Nu Plasma II MC-ICPMS at UW-Madison with a Nd:YLF-pumped Ti:sapphire femtosecond laser. These fs-LA analyses revealed samples were internally homogeneous in δ^{25} Mg. We observed a complex SIMS instrumental bias for δ^{25} Mg in olivine as a function of Mg# (100-60). The bias changed smoothly for Mg# 100-89 and 86-60 by 1.6% and 0.7%, respectively, while it jumps by 1.5% between Mg# 86 and 89. Excluding this narrow range of olivine compositions (86-89), the uncertainties of bias corrections were similar to, or smaller than, the external reproducibility of SC-Ol standards in each SIMS session.

Oxygen three-isotope analyses of AOAs were performed using the IMS 1280 with three FCs [19]. A 8 μ m diameter (0.7 nA) Cs⁺ primary ion beam was used for analyses. Synthetic pure forsterite was used as a standard and bracketed the unknown AOA analyses. The external reproducibility (2SD) of the running standard during the session was 0.4‰, 0.8‰, and 0.8‰ for δ^{18} O, δ^{17} O, and Δ^{17} O, respectively.

Results: In an oxygen three-isotope diagram, the Kaba and DOM 08006 AOAs form a tight cluster on the primitive chondrule mineral (PCM) line [9]. All AOAs measured in this study yield averaged $\delta^{18}O$ and $\Delta^{17}O$ values of $-45.8 \pm 0.7\%$ (2SD) and $-24.0 \pm 0.4\%$ (2SD), respectively.

The δ^{25} Mg values of Wild 2 olivines and AOAs from the Kaba and DOM 08006 chondrites are shown in Fig. 1. Wild 2 olivine particles show small variations in δ^{25} Mg ranging from $-0.7\pm0.3\%$ (2 σ : T175/ F1) to $0.4\pm0.4\%$ (2 σ : T149/F1). As T149/F1 and T149/F11a have Mg#'s around 86, we cannot exclude the possibility that their δ^{25} Mg values are slightly inaccurate, due to the larger instrumental bias uncertainty for the specific range of olivine compositions (Mg# 86-89). The mass independent fractionation Δ^{26} Mg values of all Wild 2 olivines, which are calibrated using the exponential law [20], are normal within analytical uncertainties ($\leq 0.8\%$: 2 σ). Variations in δ^{25} Mg are larger for AOAs (-2.6 to 1.2 %), than those of Wild 2 olivines, especially among AOAs with porous textures. Note that the DOM 08006 AOA 501 contains areas with both compact and porous textures, which show positive and negative δ^{25} Mg values, respectively.

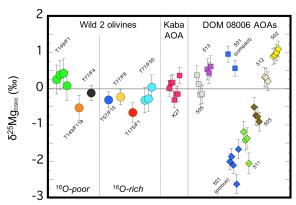


Fig. 1. δ^{25} Mg values of the Wild 2 olivine particles and meteoritic AOAs analyzed in this work. Square and diamond symbols represent compact and porous AOAs, respectively.

Discussion: All Wild 2 olivine particles have near-chondritic Mg isotope ratios, indicating there is no significant Mg isotope heterogeneity between Wild 2 olivine forming regions (¹⁶O-rich and ¹⁶O-poor) and meteorites and Earth forming regions [13].

AOAs show significant variations in δ^{25} Mg, despite their narrow ranges of 16 O-rich oxygen isotope ratios. T175/F1 and some porous AOAs have negative δ^{25} Mg, with values similar to those previously reported in other AOAs and CAIs [10-12]. This signature is indicative of condensation from nebula gas [14]. In contrast, 16 O-rich Wild 2 olivine particles, except T175/F1, do not show clear negative δ^{25} Mg values. The nearly chondritic Mg isotope ratios of 16 O-rich Wild 2 olivines could imply that: (1) Wild 2 olivines formed by nearly equilibrium condensation from neb-

ular gas; (2) after condensation from nebular gas, Wild 2 olivines experienced isotopic exchange with a chondritic component having a planetary Mg isotopic composition, and/or (3) Wild 2 olivines experienced parent body alteration. Previous studies revealed that three of the ¹⁶O-rich Wild 2 olivines (T57/F10, T77/F6, and T77/F50) have Mn-rich chemical compositions, indicating these olivines formed by condensation from nebula gas [1-6], which would support the first scenario. Note that some AOAs studied here, especially AOAs with a compact texture, show nearly chondritic or slightly heavier δ^{25} Mg values (Fig. 1). This textural and isotopic relationship suggests that compact AOAs experienced reheating processes in nebula gas. If some Wild 2 olivines experienced similar reheating processes, the second scenario may apply. Self-diffusion coefficients measured for Mg in forsterite compared to those for oxygen [see 21 and references therein], imply that the third scenario cannot be excluded at this time.

Summary: The absence of large negative δ^{25} Mg values in Wild 2 16 O-rich particles suggests either formation involved equilibrium condensation with nebula gas, or that they experienced reprocessing in nebular and/or planetary settings. Large negative δ^{25} Mg values in porous AOAs indicates a condensation origin involving kinetic mass fractionation.

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