AN EQUATION OF STATE FOR AQUEOUS SODIUM SULFATE USING LOCAL BASIS FUNCTIONS, WITH APPLICATIONS TO OCEAN WORLDS. R. James¹, S. D. Vance², and J. M. Brown³, ¹Occidental College

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Introduction: The thermodynamics fluids in ocean worlds determine how much ice is present and how the oceans transport materials and heat [1,2,3]. Sodium sulfate is identified as a possible ocean constituent on the basis of chemical models and observations [4]. Detailed models of the influence of salinity in ocean worlds rely on available thermodynamic data. To date, models of high-pressure chemistry in ocean worlds have considered only MgSO₄ [5,6]. Here, we develop a new equation of state for Na₂SO₄ incorporating available measurements of sound speed, density, and specific heat capacity.

Creating the Equation of State:

Basis Splines. We use basis splines, or b-splines, to construct a regularized fit to available sound speed measurements. Conventional equations of state use weighted polynomials to create semi-localized fits, which fail to decouple the influence of one region in P, T, and m from the behavior in other regions. B-splines solve this problem, allowing new data to be added trivially without needing to refit the original data.

Creating the Spline. Using the b-spline approach, sound speed data from Wiryana [7] and Mantegazzi [8] were fit to surfaces in P, T, and m. The global fit in P and T revealed systematic error in pressure measurements by [8], which was corrected with a constant offset in pressure along each isotherm.

Equation of State. Using the sound speed surfaces created from the data, surfaces for density, heat capacity, and Gibbs energy were created using a recursive equation of state algorithm. These surfaces were then used the Matlab implementation [9], spgft, to fit the surfaces to intermediate concentrations, and to create the comprehensive equation of state for the Na₂SO₄ (aq).

Equation of State Accuracy and Applications: Output densities as partial molal volumes match well with the available densities and partial molal volumes for Na₂SO₄. The equation of state encompasses a pressure range beyond the 1.6 GPa relevant to Ganymede, temperatures from the freezing point to 600 K and concentrations up to 1.5 mol kg⁻¹. Work in progress is formulating equations of state for crystalline salts and for water ices to more accurately and comprehensively compute phase equilibria and haline mixing in ocean worlds.

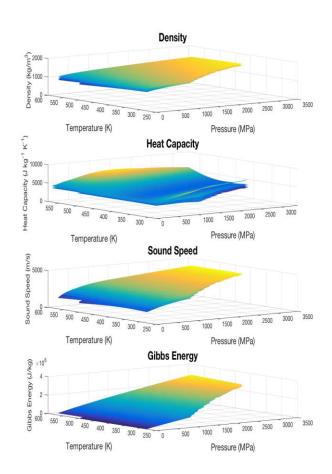


Figure 1: Density, heat capacity, sound speed, and Gibbs energy for aqueous Na₂SO₄.

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