

Shifting microbial communities sustain multi-year iron reduction and methanogenesis in ferruginous sediment incubations. M. S. Bray¹, J. Wu¹, B. C. Reed¹, C. B. Kretz², R. L. Simister³, C. Henny⁴, F. J. Stewart^{1,2}, T. J. DiChristina^{1,2}, J. A. Brandes⁵, D. A. Fowle⁶, S. A. Crowe³, J. B. Glass^{1,2#}. ¹School of Biology; ²School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA, 30332, USA. ³Departments of Microbiology & Immunology and Earth, Ocean, & Atmospheric Sciences, University of British Columbia, Vancouver, BC, Canada. ⁴Research Center for Limnology, Indonesian Institute of Sciences, Cibinong, Indonesia. ⁵Skidaway Institute of Oceanography, Savannah, GA, 31411, USA. ⁶Department of Geology, University of Kansas, Lawrence, KS, 66045, USA. #Correspondence: Jennifer.Glass@eas.gatech.edu

Introduction: Decreased solar luminosity during the Archean is thought to have been remedied by abundant greenhouse gases in the atmosphere, in particular methane (CH₄). In addition, ferruginous conditions in the Archean ocean likely led to significant deposition of solid Fe(III) to sediments. Reactive Fe(III) minerals can influence CH₄ emissions by inhibiting microbial methanogenesis or by stimulating anaerobic CH₄ oxidation. The balance between Fe(III) reduction, methanogenesis, and methane oxidation in ferruginous Archean oceans would have controlled CH₄ fluxes to the atmosphere, thereby regulating the capacity for CH₄ to warm the early Earth under the Faint Young Sun. In this study, we examined CH₄ and Fe cycling in anoxic incubations of ferruginous sediment from the Archean ocean analogue Lake Matano, Indonesia.

Study Site: Lake Matano, Indonesia is one of the only modern analogues for ferruginous Archean oceans. Despite the presence of Fe(III) oxides, CH₄ accumulates to 1.4 mM in anoxic deep waters of the lake [1]. Methanotrophy is a key carbon fixation process in Lake Matano's oxic-anoxic transition zone, and the dearth of other oxidants (<100 nM nitrate and sulfate) suggests that Fe(III) might be the primary electron acceptor in methanotrophy [2,3].

Methods: Anoxic incubations of Lake Matano sediments were sampled over three successive transfers (500 days total). Incubations contained CH₄ headspace, and 10 mM ferrihydrite or goethite. Iron reduction and methanogenesis were measured throughout the incubation by ferrozine and gas chromatography, respectively. In addition, incorporation of ¹³C-CH₄ into the DIC pool was measured as a proxy for CH₄ oxidation. Amplicon sequencing of the 16S rRNA gene was used to track microbial community changes over time.

Results: After three dilutions, Fe³⁺ reduction persisted only in bottles with ferrihydrite. Methanogenesis continued in the presence of goethite, suggesting that reactive Fe(III)-oxides (ferrihydrite) inhibit methanogenesis (Fig. 1). Methane oxidation was observed throughout the incubations, but the stoichiometry of CH₄ oxidized to Fe³⁺ reduced suggested that oxidation was linked to additional electron acceptors besides Fe³⁺. Microbial diversity decreased markedly over the course of incubation with ultimate dominance of un-

classified members of the order Desulfuromonadales in all treatments, and Rhodocyclaceae in treatments amended with CH₄.

Implications: These results suggest that Fe(III)-oxide mineralogy and availability of electron donors could have led to spatial separation of Fe(III)-reducing and methanogenic microbial communities in ancient ferruginous marine sediments. Methanogenic microbial communities could have persisted in sediments dominated by goethite or other crystalline Fe(III)-oxides while ferrihydrite would have allowed Fe(III)-reducing communities to outcompete methanogenic ones for a limited electron donor supply.

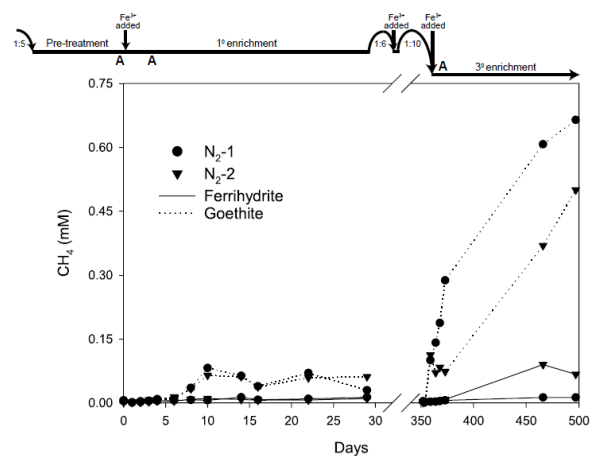


Figure 1. Accumulation of CH₄ in the headspace of sediment enrichments. Timeline at top shows transfer dates and dilution ratios. All treatments were run in duplicate.

References: [1] Crowe, S.A., et al. (2008) *Limnology and Oceanography*, 53, 319-331. [2] Crowe, S., et al., (2011) *Geobiology*, 9, 61-78. [3] Sturm, A, et al. (2016) *Biogeosciences Discussions*, 1-34

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