

Nucleobase Synthesis via UV-induced oxidation of their precursors in astrophysical ices: A Quantum Chemical Perspective. P.P.Bera¹, T. Stein², M. Head-Gordon² and T. J. Lee², NASA Ames Research Center, Space Science and Astrobiology Division, Moffett Field, CA 94035 (partha.bera@nasa.gov), ²University of California, Berkeley, USA.

Introduction: Identification of nucleobases in extra-terrestrial carbonaceous chondrites, such as Murchison, implies their formation in an abiotic condition, and supports their prebiotic role in early Earth. Physicochemical processes by which these complex molecules are synthesized in icy grains are not well understood. The products of UV photo-irradiation of purine and pyrimidine in H₂O, NH₃ and CH₄ ices have been explored using fancy new density functional theory (DFT) methods (ω B97M-V) along with large correlation consistent basis sets, and compared against laboratory experimental results. Mechanisms studied include those starting with neutral pyrimidine and purine, and their cationic counterparts, and then reacting with neutrals and radicals generated by radiation. Reaction mechanisms that involved cations on the purine or pyrimidine proved to be the ones that are most important. The calculations reveal that the formation of nucleobases is energetically and kinetically favorable. The gas phase mechanism of their formation proved ineffective, and the presence of one or several water molecules is necessary in order for the final products to form. Explicit solvent calculations using a polarized continuum model (PCM) established the effect of the ice matrix and product formation preferences. Uracil forms rather easily as oxidation is rather easy in pure H₂O ices. The scope of thymine formation in H₂O:CH₄ mixed molecular ices, however, is limited due to the inefficiency of the methylation of pyrimidine, and its oxidized derivatives. Thymine is a minor component of the products in the experimental samples. Amine group addition to purine leading to adenine and guanine in mixed NH₃ and H₂O ices is facile. Although adenine is the most likely monosubstituted photoproduct in mixed H₂O:NH₃ ice, isoguanine and xanthine are the bi-substituted products. Many of the photoproducts in UV-irradiated H₂O and pyrimidine ice mixtures are found in an experimental study. The results support the scenario in which prebiotic molecules, such as the nucleobase uracil, can be formed under abiotic processes in astrophysically relevant interstellar environments, and on surfaces of icy grains before being delivered to telluric planets such as Earth. But, it constrains the formation of thymine as well as its role in the origin of life.

References:

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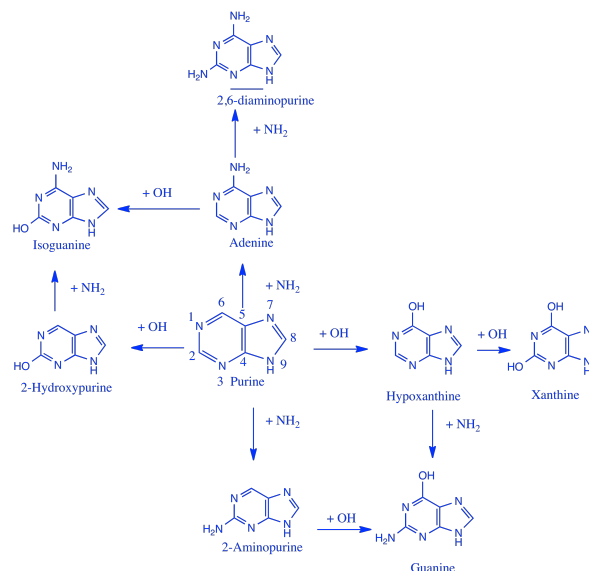


Figure 1. A reaction scheme indicating the amino and hydroxyl group substitutions on the purine ring.