

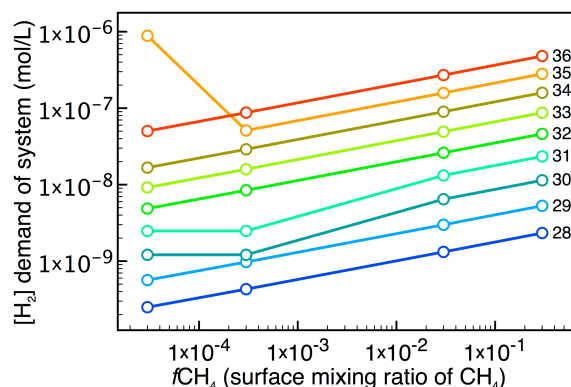
**THE THERMODYNAMICS OF LIFE ON A PLANETARY SCALE.** S. D. Domagal-Goldman<sup>1,2</sup> and S. Som<sup>2,3,4</sup>, <sup>1</sup>NASA Goddard Space Flight Center, 8800 Greenbelt Road, Greenbelt, MD 20771, USA (email: shawn.goldman@nasa.gov), <sup>2</sup>Virtual Planetary Laboratory, Seattle, WA 98125, <sup>3</sup>Blue Marble Space Institute of Science, Seattle, WA 98019, USA, <sup>4</sup>NASA Ames Research Center, Moffett Field, CA 94035, USA.

It is known that atmospheric methane ( $\text{CH}_4$ ) concentrations could have been much higher on Archean Earth. Because  $\text{CH}_4$  is an efficient greenhouse gas, this may have contributed to solutions to the faint young Sun paradox [1]. Additionally,  $\text{H}_2$  has been proposed as an important greenhouse gas for Archean Earth [2], early Mars [4], and exoplanets [5]. However, there are previously unconsidered feedbacks in this system. The conversion of  $\text{H}_2$  and  $\text{CO}_2$  (or  $\text{CO}$ ) to  $\text{CH}_4$  by biology is temperature-dependent, because of changes to the Gibbs's free energy for  $\text{CH}_4$  production from biology, and because the maintenance energy of organisms (the minimum amount of energy required by biology to fix what the environment breaks) is higher at increased temperatures [6]. We will present a study of these feedbacks that leverages a suite of models. Specifically, we couple a thermodynamic model (after 6) to calculate the relationship between surface temperature and biological  $\text{CH}_4$  production, a photochemical model (7) to predict the resulting  $\text{H}_2$  and  $\text{CH}_4$  concentrations, and a climate model (8) to predict the effects of this on surface temperature

The ecosystem model is based on the work of Hoehler (6). It utilizes Geochemists' Workbench (9) to compute dissolved equilibrium concentrations of  $\text{CO}_2$  and  $\text{CH}_4$ , which are then diffused into a single-cell bioenergetic model [10]. We use this model to predict the " $\text{H}_2$  demand" of biology to produce a certain amount of  $\text{CH}_4$  at a given temperature and dissolved inorganic carbon concentration. The photochemical model is a 1-D code that solves the mass balance equations for 74 species connected by 392 reactions. It predicts the vertical profiles of each species. This model has been previously used to simulate low- $\text{O}_2$  atmospheres, in particular those of Archean Earth [7]. We use this model to simulate the atmospheric response to biological  $\text{H}_2$  demand and  $\text{CH}_4$  production. The climate model is a 1-D radiative-convective code (8) that uses k-coefficients to simulate the radiative effects of  $\text{H}_2\text{O}$ ,  $\text{CO}_2$ , and  $\text{CH}_4$ , and includes absorption cross sections from  $\text{H}_2$  and collision-induced features of  $\text{CO}_2$ . This model predicts surface temperatures, given the atmospheric state from the photochemical model.

We will present preliminary results from this model. These will include a study of the strength of the feedbacks between  $\text{CH}_4$ , temperature, and biological productivity, and identification of stable points in the

$\text{CH}_4$ /temperature/productivity system, and resulting insights into the climate of Archean Earth. Additionally, we will discuss the implications of these feedbacks for other planets, including ancient Mars and  $\text{H}_2$ -dominated exoplanets.



**Figure 1.**  $\text{H}_2$  demand on an ecosystem to maintain different surface  $\text{CH}_4$  mixing ratios, as a function of temperature. At higher temperatures, it takes more  $\text{H}_2$  to maintain a given  $\text{CH}_4$  concentration. This creates stable points in the system (so long as a thick haze is not present). For example, increases to  $\text{CH}_4$  concentrations would lead to increased surface temperatures, but that would lead to a lower  $\text{CH}_4$  concentration given the  $\text{H}_2$  present in the system.

**References:** [1] J. Haqq-Misra, et al. (2008). *Astrobiology*, 8(6), 1127-1137. [2] Wordsworth & Pierrehumbert (2013). *Science*, 339(6115), 64-67. [3] Byrne & Goldblatt (2014). *Climate of the Past Discussions*, 10(3), 2011-2053. [4] Ramirez, et al. (2014). *Science*, 7(1), 59-63. [5] Pierrehumbert, & Gaidos (2011). *The Astrophysical Journal Letters*, 734(1), L13. [6] Hoehler, T. M. (2004). *Geobiology* 2.4 205-215. [7] Zerkle, et al. (2012). *Nature Geoscience*, 5(5), 359-363. [8] Kopparapu, et al. (2013). *The Astrophysical Journal*, 765(2), 131. [9] Bethke, C. M. (1994). *The geochemist's workbench*. University of Illinois. [10] M. J. Alperin and T. M. Hoehler. (2009). *American Journal of Science* 309.10: 869-957.